

# Taste and Understand Acid Additions

## I. History

### A. Robert Boyle

1. Acids: Taste sour, corrode metal, change litmus paper red, becomes less acid when mixed with bases.
2. Bases: Feel slippery, change litmus paper blue, becomes less basic when mixed with acids.

### B. Svante Arrhenius

1. Water can dissolve compounds
  - a. Acids give off a hydrogen ion ( $H^+$ )
  - b. Bases give off a hydroxide ion ( $OH^-$ )
2. Neutralization
  - a. Water
  - b. Salt

### C. Johannes Bronsted and Thomas Lowry

1. Acid: Any substance that gives off a hydrogen ion
2. Base: Any substance that can accept a hydrogen ion

## II. Acid

### A. Grapes

1. Tartaric Acid
  - a. Most abundant acid in grapes
  - b. Partially consumed during primary fermentation
2. Malic Acid
  - a. Second most abundant acid in grapes
  - b. Mostly consumed during malolactic fermentation
  - c. Converted to Lactic acid during malolactic fermentation
3. Citric Acid
  - a. Trace amount, mostly used for internal functions of the grape
  - b. Broken down and consumed during malolactic fermentation once Malate is gone

### B. Titratable Acidity

1. Measures the total amount of acid in the grape juice / wine (g/L)
2. Effects the overall taste and composition of the wine

### III. pH

- A. Effects growth of undesirable bacteria and yeast
- B. Greater sulfur dioxide anti-oxidant/anti-microbial activity
- C. Impacts oxidation rates
- D. Favors longer ageing potential
- E. Measures the total amount of hydrogen ions (H<sup>+</sup>)
- F.  $\text{pH} = -\log[\text{H}^+]$

Strong Acid	1.00E+00	1.0	0	0.000000000000001	1.00E-14
	1.00E-01	0.1	1	0.00000000000001	1.00E-13
	1.00E-02	0.01	2	0.0000000000001	1.00E-12
	1.00E-03	0.001	3	0.000000000001	1.00E-11
	1.00E-04	0.0001	4	0.0000000001	1.00E-10
Weak Acid	1.00E-05	0.00001	5	0.000000001	1.00E-09
	1.00E-06	0.000001	6	0.00000001	1.00E-08
Neutral	1.00E-07	0.0000001	7	0.0000001	1.00E-07
	1.00E-08	0.00000001	8	0.000001	1.00E-06
Weak Base	1.00E-09	0.000000001	9	0.00001	1.00E-05
	1.00E-10	0.0000000001	10	0.0001	1.00E-04
	1.00E-11	0.00000000001	11	0.001	1.00E-03
	1.00E-12	0.000000000001	12	0.01	1.00E-02
	1.00E-13	0.0000000000001	13	0.1	1.00E-01
Strong Base	1.00E-14	0.00000000000001	14	1.0	1.00E+00

## Base

14	Drain-O & Sodium Hydroxide
13.8	Oven Cleaner
13.05	Bottle washing (process midpoint)
12.6	Bleach
12.4	Lime, saturated
12	Photographic Developer
11.6	Ammonia, N
10	Milk of Magnesia & Great Salt Lake
9.2	Borax
8.4	Sodium bicarbonate
8	Egg white & Sea water
7.3-7.5	Blood, Human
7	Pure water
6	Milk
5	Cottage cheese & Black Coffee
4.0-5.0	Beer
4	Acid Rain & Tomatoes
3.0-4.0	Oranges
3.5	Pickling
2.8-3.8	Wines
3	Coca-Cola
2.8-3.0	Vinegar
2	Limes & Lemons
0.8	Hydrochloric (Stomach) Acid
0.3	Sulfuric, Nitric, & Battery Acids

## Acid